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CONCERNING A FILING UNDER 35 U.S.C. 371INTERNATIONAL APPLICATION NO.  
PCT/JP00/00607INTERNATIONAL FILING DATE  
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## TITLE OF INVENTION

## PROCESS FOR PREPARING BATTERY AND BATTERY USING SAME

APPLICANT(S) FOR DO/EO/US  
HIROI Osamu et al.

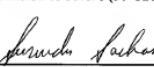
Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

- This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
- This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
- This is an express request to begin national examination procedures (35 U.S.C. 371(f)). The submission must include items (5), (6), (9) and (24) indicated below.
- The US has been elected by the expiration of 19 months from the priority date (Article 31).
- A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
  - is attached hereto (required only if not communicated by the International Bureau).
  - has been communicated by the International Bureau.
  - is not required, as the application was filed in the United States Receiving Office (RO/US).
- An English language translation of the International Application as filed (35 U.S.C. 371(c)(2))
  - is attached hereto.
  - has been previously submitted under 35 U.S.C. 154(d)(4).
- Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
  - are attached hereto (required only if not communicated by the International Bureau).
  - have been communicated by the International Bureau.
  - have not been made; however, the time limit for making such amendments has NOT expired.
  - have not been made and will not be made.
- An English language translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
- An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
- An English language translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).
- A copy of the International Preliminary Examination Report (PCT/IPEA/409).
- A copy of the International Search Report (PCT/ISA/210).

## Items 13 to 20 below concern document(s) or information included:

- An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
- An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
- A **FIRST** preliminary amendment.
- A **SECOND** or **SUBSEQUENT** preliminary amendment.
- A substitute specification.
- A change of power of attorney and/or address letter.
- A computer-readable form of the sequence listing in accordance with PCT Rule 13ter.2 and 35 U.S.C. 1.821 - 1.825.
- A second copy of the published international application under 35 U.S.C. 154(d)(4).
- A second copy of the English language translation of the international application under 35 U.S.C. 154(d)(4).
- Certificate of Mailing by Express Mail
- Other items or information:

Request for Consideration of Documents Cited in International Search Report  
Drawings (2 Sheets)

U.S. APPLICATION NO. (IF KNOWN - SEE 37 CFR <b>97 926264</b>	INTERNATIONAL APPLICATION NO. <b>PCT/JP00/00607</b>	ATTORNEY'S DOCKET NUMBER <b>210245US2PCT</b>						
24. The following fees are submitted: <b>BASIC NATIONAL FEE (37 CFR 1.492 (a) (1) - (5)) :</b>		<b>CALCULATIONS PTO USE ONLY</b>						
<input type="checkbox"/> Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO .....		<b>\$1040.00</b>						
<input checked="" type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO .....		<b>\$890.00</b>						
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<b>ENTER APPROPRIATE BASIC FEE AMOUNT = \$890.00</b>								
Surcharge of <b>\$130.00</b> for furnishing the oath or declaration later than months from the earliest claimed priority date (37 CFR 1.492 (e)).								
<b>\$20      \$30      \$0.00</b>								
CLAIMS	NUMBER FILED	NUMBER EXTRA						
Total claims	12 - 20 =	0						
Independent claims	2 - 3 =	0						
Multiple Dependent Claims (check if applicable).		<input type="checkbox"/>						
<b>TOTAL OF ABOVE CALCULATIONS = \$890.00</b>								
<input type="checkbox"/> Applicant claims small entity status. See 37 CFR 1.27). The fees indicated above are reduced by 1/2.								
<b>\$0.00</b>								
<b>SUBTOTAL = \$890.00</b>								
Processing fee of <b>\$130.00</b> for furnishing the English translation later than months from the earliest claimed priority date (37 CFR 1.492 (f)).								
<b>\$20      \$30      \$0.00</b>								
<b>TOTAL NATIONAL FEE = \$890.00</b>								
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable).								
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<b>TOTAL FEES ENCLOSED = \$890.00</b>								
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a. <input checked="" type="checkbox"/> A check in the amount of <b>\$890.00</b> to cover the above fees is enclosed. b. <input type="checkbox"/> Please charge my Deposit Account No. _____ in the amount of _____ to cover the above fees. A duplicate copy of this sheet is enclosed. c. <input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit Account No. <b>15-0030</b> A duplicate copy of this sheet is enclosed. d. <input type="checkbox"/> Fees are to be charged to a credit card. <b>WARNING: Information on this form may become public. Credit card information should not be included on this form.</b> Provide credit card information and authorization on PTO-2038.								
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SEND ALL CORRESPONDENCE TO:								
 <b>22850</b> Surinder Sachar Registration No. 34,423								
 <b>SIGNATURE</b> Marvin J. Spivak <b>NAME</b> <b>24,913</b> <b>REGISTRATION NUMBER</b> <b>Oct. 3 2001</b> <b>DATE</b>								

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## DESCRIPTION

PROCESS FOR PREPARING BATTERY AND BATTERY USING SAME

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## TECHNICAL FIELD

The present invention relates to a process for preparing a battery. More particularly, the present invention relates to a process for preparing a lithium secondary battery.

10

## BACKGROUND ART

Presently, lithium ion secondary batteries are being actively upgraded as a suitable secondary battery for achieving high voltage and high energy density. Major components thereof are a rolled out electrode comprising an electrode which has a pair of positive and negative electrodes, and a separator which separates each electrode for preventing short-circuit, and an electrolytic solution filled to the rolled out electrode. Practically used lithium secondary batteries are prepared according to procedures comprising laminating and winding the both of the positive, negative electrodes together with a separator to be packaged, filling an electrolytic solution and sealing the battery. Examples of these preparation processes are found in Japanese Unexamined Patent Publication No. 334884/1998 and the like.

An ideal amount of an electrolytic solution is such that voids in the porous positive, negative electrodes and separator are suitably filled with the electrolytic solution. The above procedures, however, cause the electrolytic solution to be filled into spaces irrelevant to the function of a battery, for example, into voids between the outer can and

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the electrode, and into the gap of the roll formed by lamination winding of the electrode. Since the electrolytic solution is present even in the spaces irrelevant to the battery function in this way, the electrolytic solution accounts for remarkably more than a desired amount, being the 5 reason of weight increase.

The electrolytic solution present in the spaces irrelevant to the battery function flows and leaks more easily compared to the electrolytic solution retained in the pores of the electrodes and the separator. Therefore, it can be the reason for troubles with liquid leak 10 of the battery.

As mentioned above, it is preferable to lessen the amount of the electrolytic solution in the spaces irrelevant to the battery function from the viewpoints of weight-wise as well as preventing the battery liquid from leaking.

15

#### DISCLOSURE OF INVENTION

The present invention has been carried out in order to solve the above problems. The object of the present invention is to provide a battery which has an extremely small amount of electrolytic 20 solution in the above spaces irrelevant to the battery function. The present inventors have studied on processes for preparing a battery in an effort to obtain a downsized, lightweight battery with high performance, and completed the following process and device for preparing a battery as well as a battery.

25

The present invention relates to:  
a process for preparing a battery comprising,  
(i) filling an electrolytic solution to a rolled out electrode,

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(ii) removing part of the electrolytic solution on the rolled out electrode to which the electrolytic solution is filled, and

(iii) packaging the rolled out electrode from which the electrolytic solution is partly removed (Claim 1);

5 the process for preparing a battery of Claim 1, wherein the rolled out electrode is immersed in the electrolytic solution to fill the electrolytic solution thereto (Claim 2);

the process for preparing a battery of Claim 1, wherein part of the electrolytic solution on the rolled out electrode is removed under air

10 current (Claim 3);

the process for preparing a battery of Claim 3, wherein jetting air through a nozzle is applied to the rolled out electrode (Claim 4);

the process for preparing a battery of Claim 4, wherein the jetting air-applied positions of the rolled out electrode are changed (Claim 5);

15 the process for preparing a battery of Claim 3, wherein air current is applied to the rolled out electrode by depressurizing one side of a flow channel in which the rolled out electrode is inserted (Claim 6);

the process for preparing a battery of Claim 3, wherein air current is applied to the rolled out electrode by pressurizing one side of a flow

20 channel in which the rolled out electrode is inserted (Claim 7);

the process for preparing a battery of Claim 1, wherein part of the electrolytic solution on the rolled out electrode is removed by contacting thereto an object capable of absorbing and retaining the electrolytic solution (Claim 8);

25 the process for preparing a battery of Claim 1, further comprising, diluting part of the electrolytic solution on the rolled out electrode by using a solvent which is mutually soluble with the electrolytic solution

(Claim 9);

the process for preparing a battery of Claim 9, wherein a solvent for the electrolytic solution comprises the solvent which is mutually soluble with the electrolytic solution (Claim 10);

5 a device for preparing a battery, comprising  
(i) means for filling an electrolytic solution to a rolled out electrode,  
(ii) means for removing part of the electrolytic solution on the rolled out electrode to which the electrolytic solution is filled, and  
(iii) means for packaging the rolled out electrode from which the  
10 electrolytic solution is partly removed (Claim 11); and  
a battery prepared according to the process for preparing a battery of  
Claim 1 (Claim 12).

#### BRIEF DESCRIPTION OF DRAWINGS

15 Fig. 1 is a schematic view outlining the step for removing part of an electrolytic solution on a rolled out electrode by applying a jetting air through nozzles according to one embodiment of the present invention.

20 Fig. 2 is a schematic view outlining the shape of a nozzle having a slit exhaust port according to one embodiment of the present invention.

Fig. 3 is a schematic view outlining the shape of a nozzle having an exhaust port with array of holes according to one embodiment of the present invention.

25 Fig. 4 is a schematic view outlining the step for removing part of an electrolytic solution on a rolled out electrode by means of air current generated by depressurizing one side of a flow channel into

which the rolled out electrode is disposed.

In Fig. 1 numeral 1 indicates rolled out electrode, numeral 2 indicates a nozzle which jets gas out and numeral 3 indicates jetting air.

In Fig. 2 numeral 2 indicates a nozzle which jets gas out and 5 numeral 3 indicates jetting air.

In Fig. 3 numeral 2 indicates a nozzle which jets gas out and numeral 3 indicates jetting air.

In Fig. 4 numeral 1 indicates a rolled out electrode, numeral 3 indicates air current and numeral 4 indicates a flow channel of the air

10 current.

#### BEST MODE FOR CARRYING OUT THE INVENTION

The embodiments of the present invention are explained below.

15 The process for preparing a battery of the present invention comprises (i) filling an electrolytic solution to a rolled out electrode obtained by unifying a positive electrode, a negative electrode and a separator, (ii) removing part of the electrolytic solution on the rolled out electrode to which the electrolytic solution is filled and (iii) packaging 20 the rolled out electrode from which the electrolytic solution is partly removed. The steps are explained in detail below.

25 The rolled out electrode of the battery is obtained by unifying a positive electrode, a negative electrode and a separator including, for example, those prepared by lamination and winding of a positive electrode sheet, a negative electrode sheet and a separator film for separation thereof. Adhesion of laminated, wound elements by a resin or the like, or fixation on the exterior with tape or the like brings about a

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rolled out electrode as an independent structure.

Useful as the positive electrode is, for example, the one obtained by applying powdery lithium cobaltate onto a piece of aluminum foil using poly(vinylidene fluoride) as a binder and by 5 drying. Useful as the negative electrode is, for example, the one obtained by applying powdery graphite carbon onto a piece of copper foil using poly(vinylidene fluoride) as a binder and by drying. A porous polyethylene film can be used as the separator.

A battery comes to function when the rolled out electrode, 10 the independent structure, is charged with an electrolytic solution. As the electrolytic solution, for example, the one obtained by dissolving an electrolyte into any solvent can be used. The useful solvent for the electrolytic solution includes, for example, a mixed solvent of ethylene carbonate and dimethyl carbonate. Examples of the electrolyte are 15 lithium hexafluorophosphate and the like.

Filling of the electrolytic solution can be performed outside the outer package of the battery. It is difficult to immerse and fill the electrolytic solution efficiently to the innermost of the rolled out electrode when the rolled out electrode is relatively large or when a 20 dense layer exists in the electrode. The electrolytic solution can be efficiently immersed and filled thereto at a high filling rate by depressurizing or centrifuging the rolled out electrode in a state of complete sinking beneath the electrolytic solution. The depressurization pressure can be, for example, a little higher than the 25 pressure which causes depressurization boiling of the electrolytic solution. Centrifugal force can be, for example, such that rotational speed of 4 G, about four times the gravity, is thereby generated.

When the rolled out electrode filled with the electrolytic solution is taken out from the electrolytic solution, it holds some electrolytic solution irrelevant to the battery function (extra electrolytic solution) on the outside or in a gap of the roll. It is

5 preferable to decrease the extra electrolytic solution as much as possible since it causes weight increase and liquid leak.

Partial removal of the electrolytic solution on the rolled out electrode to which the electrolytic solution is filled enables to decrease the amount of the extra electrolytic solution present in the

10 spaces irrelevant to the battery function. Part of the electrolytic solution on the rolled out electrode can be removed by applying air current to the rolled out electrode. In case of the partial removal of the electrolytic solution on the rolled out electrode under air current, dry air or dry nitrogen, for example, is available as a gas which

15 generates the air current. The dry air and the dry nitrogen do not harm the battery performances.

Processes for removing the extra electrolytic solution includes a process for spraying fast air current. The application of jetting air through the nozzle to the rolled out electrode enables to

20 remove part of the electrolytic solution on the rolled out electrode. An outline of the step for removing part of an electrolytic solution on a rolled out electrode by applying the jetting air through nozzles is shown in Fig. 1. The spraying of the jetting air through the nozzles to the surface of the rolled out electrode results in removal of the

25 electrolytic solution thereon by the pressure.

It is possible to remove the extra electrolytic solution efficiently all over the rolled out electrode by changing (scanning) the

position at which removal strength of the electrolytic solution is high, namely, the position received the direct air current in application of the jetting air through the nozzle to the rolled out electrode. Scanning for the positions received the direct air current can be

5 performed by moving the rolled out electrode or the nozzle in application of the jetting air through the nozzle to the rolled out electrode. The scanning rate can be, for example, about 5 cm per second.

Preferably, the jetting air through the nozzle ranges in  
10 angle from 30 to 90° to the normal line of the electrolytic solution-removed plane of the rolled out electrode. When it is less than 30°, the droplets do not disperse in one direction causing to increase the possibility of polluting the surface from which the electrolytic solution has already been removed.

15 Preferable as a nozzle are those having a slit exhaust port which jets air out or an exhaust port which jets air out through array of holes for sending out band-shaped air current. An outline of the nozzle having a slit exhaust port is shown in Fig. 2. An outline of the nozzle having an exhaust port with array of holes is shown in Fig. 3.

20 The jetting out rate of the air current through the nozzle can be, for example, about 40 m per second. When the jetting out rate of the air current through the nozzle is too slow sufficient removal of the extra electrolytic solution tends to be difficult. When it is too fast the rolled out electrode may be deformed by means of the  
25 air current pressure.

Processes for removing the extra electrolytic solution includes a process for applying fast air current through a flow

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channel to the rolled out electrode being disposed inside the flow channel. The flow rate of the air current through the flow channel can be, for example, 20 m per second. When the flow is too fast, the rolled out electrode tends to be deformed. When it is too slow, 5 sufficient removal of the extra electrolytic solution tends to be difficult.

The air current can be formed by sending gas from one side of the flow channel or by depressurizing one side of the flow channel with the rolled out electrode being disposed therein. An outline of 10 the step for removing part of an electrolytic solution on a rolled out electrode by means of air current generated by depressurizing one side of a flow channel is shown in Fig. 4.

Processes for removing the extra electrolytic solution includes a process for contacting an object capable of absorbing the 15 electrolytic solution. Useful as the object capable of absorbing and retaining the electrolytic solution are, for example, a liquid-absorbent non-woven fabric, woven fabric, sponge resin and the like.

Packaging the rolled out electrode from which part of the applied electrolytic solution is removed enables to prepare a battery, 20 wherein the extra electrolytic solution irrelevant to the battery function is remarkably reduced. Useful as the outer can is, for example, a metal can made of stainless or aluminum or an aluminum laminated film.

According to the present invention, lightweight of batteries can be achieved and possibility of liquid leak can be remarkably 25 decreased by reducing the extra electrolytic solution.

In some case, solid content in an electrolytic solution may remarkably precipitate onto the surface of a rolled out electrode in

accordance with vaporization of a solvent through the removal process of the extra electrolytic solution, depending on kinds of electrolytic solution. Preparation of batteries by using the rolled out electrode on which the solid content is precipitated may cause

5 unevenness of the outer can surface, and may result in unfavorable appearance.

Precipitation of the solid content can be reduced by diluting part of the electrolytic solution on the rolled out electrode by using a solvent (cleaning solvent) which has less solid content than the

10 electrolytic solution and is mutually soluble with components of the electrolytic solution. For example, part of the electrolytic solution on the rolled out electrode can be diluted by washing the rolled out electrode with the cleaning solvent.

Available as the cleaning solvent are solvents in which no or

15 little solid content is dissolved including a solvent having less solid content than the electrolytic solution, while it is also possible to use a solvent which does not affect the battery performances or which is mutually soluble with the electrolytic solution. For example, a solvent obtained by removing a solute such as salt from the components of the

20 electrolytic solution can be preferably used since it does not affect battery performances. Examples of the useful cleaning solvent are diethyl carbonate, dimethyl carbonate, methyl ethyl carbonate, propylene carbonate, caprolactone and the like.

Washing of the rolled out electrode with these kinds of

25 solvent before or after the removal step of the extra electrolytic solution can prevent the solid content from precipitating.

According to the process for preparing batteries of the

present invention, the electrolytic solution irrelevant to the battery performances is decreased, making it possible to prepare a battery which is light, hard to leak liquid and has no problems with battery performances or appearance.

5 The device for preparing batteries according to the present invention comprises (i) means for filling an electrolytic solution to a rolled out electrode obtained by unifying a positive electrode, a negative electrode and a separator, (ii) means for removing part of the electrolytic solution on the rolled out electrode to which the electrolytic solution is  
10 filled, and  
(iii) means for packaging the rolled out electrode from which the electrolytic solution is partly removed.

15 The means for filling an electrolytic solution to a rolled out electrode has, for example, a tank for electrolytic solution used for immersion of the rolled out electrode in the electrolytic solution. The means for removing part of the electrolytic solution on the rolled out electrode has, for example, nozzles jetting gas out, a flow channel and a fan or depressurization device for forming air current, or a substance capable of absorbing the electrolytic solution.

20

#### EXAMPLES

While the present invention is explained in detail through examples below, the present invention is not limited thereto.

25

#### EXAMPLE 1

(Preparation of electrolytic solution)

Powdery LiPF<sub>6</sub> as a support salt was dissolved into a solvent

obtained by mixing ethylene carbonate and dimethyl carbonate in a weight ratio of 1:1 to adjust it to 1 mol/l.

(Preparation of electrodes)

Positive electrode material was prepared by applying a paste 5 of a positive electrode active material obtained by mixing 87 % by weight of an active material,  $\text{LiCoO}_2$ , 8 % by weight of a conductive graphite powder, 5 % by weight of poly(vinylidene fluoride) as a binder resin and a solvent, N-methylpyrrolidone (hereinafter referred to as "NMP") onto a current terminal comprising a piece of 20- $\mu\text{m}$ -thick aluminum foil 10 according to Doctor Blade method in a thickness of about 200  $\mu\text{m}$ , drying the same, and further, by rolling it to 120  $\mu\text{m}$  thick.

A negative electrode material was prepared by applying and drying a paste of a negative electrode active material obtained by mixing 95 % by weight of mesophase microbeads carbon (available from Osaka 15 Gas Co., Ltd.), 5 % by weight of poly(vinylidene fluoride) as a binder resin and a solvent, NMP onto a current terminal comprising a piece of 12- $\mu\text{m}$ -thick copper foil according to Doctor Blade method in a thickness of about 200  $\mu\text{m}$ , and further, rolling it to 120  $\mu\text{m}$  thick.

The thus prepared positive and negative electrode materials 20 were cut into a size of 50 mm  $\times$  200 mm and current collector terminals were attached to the ends.

(Formation of rolled out electrode)

A separator cut into a size of 52 mm  $\times$  210 mm (available from 25 Höchst Celanese Co., Ltd.; Trade-name: CELLGUARD#2400) was interposed between the positive electrode and the negative electrode. It was rolled out with the lamination maintained, flattened and fixed with a strip of adhesive tape to obtain a plane, wound construction. The

number of the rolling was adjusted so that the width is about 50 mm when flattened.

(Injection of electrolytic solution)

A rolled out electrode was disposed in an electrolytic solution.

5 The whole was depressurized with a pump and maintained at 50 torr for 3 minutes, and then the atmospheric pressure was recovered.

(Removal of extra electrolytic solution)

The electrolytic solution was removed by applying dry jetting air through 70-mm-long and 0.3-mm-wide slit nozzles in one liter per second in terms of the atmospheric pressure. The jetting air was applied at a slope of 45° to the normal line of the rolled out electrode surface while distance from the nozzle tip to the rolled out electrode surface was set to 5 mm. Removal was performed by scanning the rolled out electrode from one end to the other end at a speed of 5 cm per second. The both surfaces of the rolled out electrode underwent the above procedure.

A very small amount of solid precipitation was observed on the rolled out electrode surface after the removal procedure of the extra electrolytic solution.

20 (Outer packaging using aluminum laminated film)

The rolled out electrode prepared through the above procedure was kept under reduced pressure of about 50 torr and an aluminum laminated film was used for sealing to form an outer package.

25 A piece of 50- $\mu$ m-thick aluminum foil, 12- $\mu$ m-thick polyethylene terephthalate film and 5- $\mu$ m-thick polyethylene film were laminated, and it was cut into a size of 70 mm  $\times$  120 mm to be used as an aluminum laminated film. The film was folded in half, 70 mm  $\times$  60 mm,

and three remaining edges were closed by heat-sealing with the rolled out electrode therein.

(Evaluation of liquid leak)

5 A hole having a diameter of 2 mm was made at the edge of the outer package and the package was kept with the hole downward for 5 minutes to estimate the amount of the leaked electrolytic solution according to the weight change before-and-after the keeping.

10 The battery prepared through the above process was 13.5 g in weight and showed 0.05 g of liquid leak according to the liquid leak evaluation.

EXAMPLE 2

(Case of disposition in flow channel)

15 There was provided a cylindrical flow channel in which 0.5 mm of space is formed between the flow channel inner wall and the rolled out electrode when the battery rolled out electrode was placed inside. The rolled out electrode was fixed on the inside of the flow channel to remove the extra electrolytic solution by applying dry air current to the space between the flow channel and the rolled out 20 electrode. The amount of the dry air was adjusted so that the pressure difference was 0.5 kgf/cm<sup>2</sup> between the upper side and the lower side.

25 A battery was prepared in the same manner as in Example 1 except that the extra electrolytic solution was removed according to the above process. The battery was 13.3 g in weight and showed 0.01 g of liquid leak according to the liquid leak evaluation.

### EXAMPLE 3

(Case of using non-woven fabric)

A battery was prepared in the same manner as in Example 1 except that the extra electrolytic solution was removed by using a 5 polypropylene non-woven fabric to wipe out the electrolytic solution on the outside of the post-injection rolled out electrode. The battery was 13.6 g in weight and showed 0.07 g of liquid leak according to the liquid leak evaluation. Though this process does not require any device for supplying a large amount of dry gas, it was impossible to remove the 10 extra solution in the gap of the roll.

### EXAMPLE 4

(Case of washing by solvent)

A battery was prepared in the same manner as in Example 1 except that removal of the extra electrolytic solution was performed after 15 the rolled out electrode posterior to filling of the electrolytic solution was immediately immersed in diethyl carbonate for 10 seconds.

The battery was 13.5 g in weight and showed 0.05 g of liquid leak according to the liquid leak evaluation. Such a small solid 20 precipitation on the rolled out electrode surface as seen in Example 1 was not observed. No unevenness according to solid precipitation surfaced even after the packaging.

### COMPARATIVE EXAMPLE 1

An aluminum laminated film was sealed remaining one edge 25 unsealed and it was made into a bag with the rolled out electrode interposed therebetween. An electrolytic solution was injected into the

bag made of the package film to perform immersion of the electrolytic solution under reduced pressure. The extra electrolytic solution after the immersion was removed by pipette and then, the remaining edge was sealed to close. A battery was prepared in the same manner as in  
5 Example 1 except for the above procedures. The battery was 14.1 g in weight and showed 0.9 g of liquid leak according to the liquid leak evaluation.

As shown in this Comparative Example, removal of the extra electrolytic solution after packaging makes it difficult to reduce the extra  
10 electrolytic solution not contributory to the battery performance sufficiently. Consequently, the battery will weigh heavier while the electrolytic solution causing liquid leak will be increased.

#### COMPARATIVE EXAMPLE 2

15 An aluminum laminated film was sealed remaining one edge unsealed, and it was made into a bag with the rolled out electrode interposed therebetween. An electrolytic solution was injected into the bag made of the package film and the total weight of the battery was adjusted to 13.5 g. The weight is the same as that in Example 1. After  
20 that it was kept under 50 torr for 3 minutes followed by sealing of one remaining edge to close the battery. A battery was prepared in the same manner as in Example 1 except for the above procedures.

The battery was 13.5 g in weight as mentioned above showing  
0.07 g of liquid leak according to the liquid leak evaluation. When  
25 charge-discharge tests were carried out by using this battery and the battery of Example 1 at the rate of 0.5 C, this battery accounted for about 80 % the discharge capacitance compared with the battery

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prepared in Example 1.

This is resulted from the fact that the electrolytic solution did not sufficiently fill the spaces where the electrolytic solution is needed in terms of battery performances, being present even in spaces not contributory to the battery performances.

#### EXAMPLE 5

(Case of slow air current)

A battery was prepared in the same manner as in Example 1 except that the air current was applied at a rate of 0.2 liter per second when the extra electrolytic solution was removed. Even after the removal of the extra electrolytic solution, remainder of electrolytic solution was remarkably found on the surface of the rolled out electrode. The battery was 13.9 g in weight showing 0.5 g of liquid leak according to the liquid leak evaluation. Although there were weight increase due to the extra electrolytic solution and increase of electrolytic solution leading to liquid leak compared with Example 1, remarkable effect was seen in comparison with Comparative Example 1.

#### 20 COMPARATIVE EXAMPLE 4

(Case of fast air current)

A battery was prepared in the same manner as in Example 1 except that the air current was applied at a rate of 2 liter per second when the extra electrolytic solution was removed.

25 Peeling was partly observed among components of the rolled out electrode after the removal of the extra electrolytic solution. The battery was 13.4 g in weight showing 0.02 g of liquid leak according to

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the liquid leak evaluation.

According to the preparation process of Claims 1 to 8, it is possible to prepare a light battery with less possibility of liquid leak since the amount of the extra electrolytic solution not contributory to the 5 battery performance is small.

According to the preparation process of Claims 9 to 10, it is possible to prepare a battery having little, visually unfavorable unevenness which is caused by precipitation on the surface of the rolled out electrode.

10 According to the preparation device of Claim 11, it is possible to prepare a light battery with less possibility of liquid leak.

The battery according to Claim 12 has high weight energy density because the weight of the electrolytic solution is reduced and it has small possibility of liquid leak accident because the extra electrolytic 15 solution is decreased.

#### INDUSTRIAL APPLICABILITY

The process for preparing a battery, the device for preparing a battery and the battery of the present invention can be 20 widely applied to a battery including lithium ion battery which comprises a rolled out electrode immersed in an electrolytic solution and an outer can packaging the same.

CLAIMS

1. A process for preparing a battery comprising,  
5 (i) filling an electrolytic solution to a rolled out electrode obtained by  
unifying a positive electrode, a negative electrode and a separator,  
(ii) removing part of the electrolytic solution on the rolled out electrode to  
which the electrolytic solution is filled, and  
(iii) packaging the rolled out electrode from which the electrolytic  
solution is partly removed.

10 2. The process for preparing a battery of Claim 1, wherein the  
rolled out electrode is immersed in the electrolytic solution to fill the  
electrolytic solution thereto.

15 3. The process for preparing a battery of Claim 1, wherein  
part of the electrolytic solution on the rolled out electrode is removed  
under air current

20 4. The process for preparing a battery of Claim 3, wherein the  
jetting air through a nozzle is applied to the rolled out electrode.

25 5. The process for preparing a battery of Claim 4, wherein the  
jetting air-applied positions of the rolled out electrode are changed.

6. The process for preparing a battery of Claim 3, wherein air  
current is applied to the rolled out electrode by depressurizing one side  
of a flow channel in which the rolled out electrode is inserted.

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7. The process for preparing a battery of Claim 3,  
wherein air current is applied to the rolled out electrode by pressurizing  
one side of a flow channel in which the rolled out electrode is inserted.

5           8. The process for preparing a battery of Claim 1,  
wherein part of the electrolytic solution on the rolled out electrode is  
removed by contacting thereto an object capable of absorbing and  
retaining the electrolytic solution.

10           9. The process for preparing a battery of Claim 1, further  
comprising, diluting part of the electrolytic solution on the rolled out  
electrode by using a solvent which mutually soluble with the electrolytic  
solution.

15           10. The process for preparing a battery of Claim 9,  
wherein a solvent for the electrolytic solution comprises the solvent  
which is mutually soluble with the electrolytic solution.

11. A device for preparing a battery, comprising  
20 (i) means for filling an electrolytic solution to a rolled out electrode  
obtained by unifying a positive electrode, a negative electrode and a  
separator, and  
(ii) means for removing part of the electrolytic solution on the rolled out  
electrode to which the electrolytic solution is filled.

25           12. A battery prepared according to the process for  
preparing a battery of Claim 1.

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FIG. 1

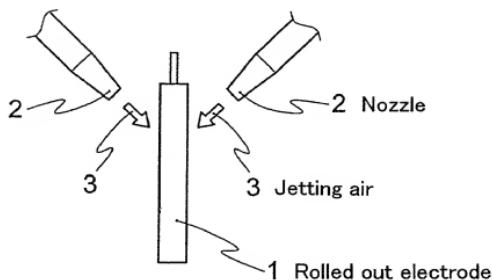


FIG. 2

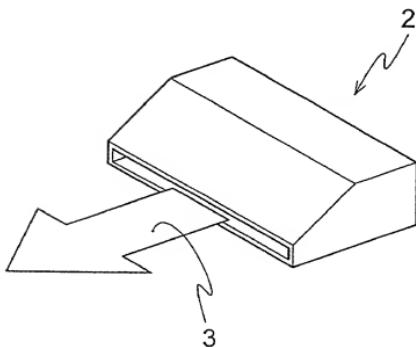


FIG. 3

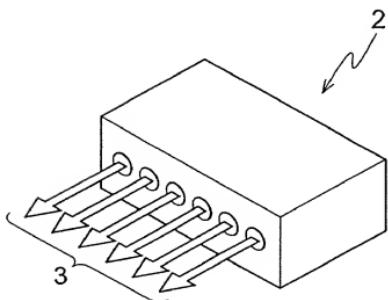
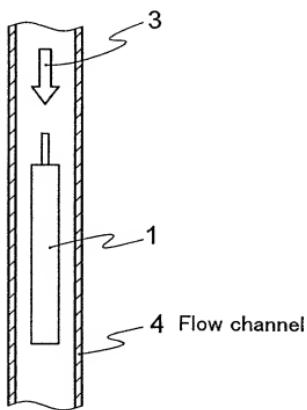


FIG. 4



## Declaration and Power of Attorney For Patent Application

### 特許出願宣言書及び委任状

## Japanese Language Declaration

## 日本語宣言書

下記の氏名の発明者として、私は以下の通り宣言します。

As a below named inventor, I hereby declare that:

私の住所、私書箱、国籍は下記の私の氏名の後に記載された通りです。

My residence, post office address and citizenship are as stated next to my name.

下記の名称の発明に関する請求範囲に記載され、特許出願している発明内容について、私が最初かつ唯一の発明者（下記の氏名が一つの場合）もしくは最初かつ共同発明者（下記の名称が複数の場合）であると信じています。

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

PROCESS FOR PREPARING BATTERY

## AND BATTERY USING SAME

the specification of which

is attached hereto.

was filed on February 4, 2000.

as United States Application Number or

PCT International Application Number

(if applicable)

私は、特許請求範囲を含む上記訂正後の明細書を検討し、内容を理解していることをここに表明します。

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

私は、連邦規則法典第37編第1条56項に定義されるとおり、特許資格の有無について重要な情報を開示する義務があることを認めます。

I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, Section 1.56.

## Japanese Language Declaration

(日本語宣言書)

私は、米国法典第35編119条 (a) - (d) 項又は365条 (b) 項に基づき下記の、米国以外の国の少なくとも一ヵ国を指定している特許協力条約365 (a) 項に基づく国際出願、又は外国での特許出願もしくは発明者証の出願についての外国優先権をここに主張するとともに、優先権を主張している、本出願の前に出願された特許または発明者証の外国出願を以下に、枠内をマークすることで、示しています。

## Prior Foreign Application(s)

外国での先行出願

(Number) (番号)	(Country) (国名)	(Day/Month/Year Filed) (出願年月日)	Priority Claimed 優先権主張
(Number) (番号)	(Country) (国名)	(Day/Month/Year Filed) (出願年月日)	Yes はい <input type="checkbox"/> No いいえ <input type="checkbox"/>

私は、第35編米国法典119条 (e) 項に基づいて下記の米国特許出願規定に記載された権利をここに主張いたします。

(Application No.) (出願番号)	(Filing Date) (出願日)	(Application No.) (出願番号)	(Filing Date) (出願日)
-----------------------------	------------------------	-----------------------------	------------------------

私は、下記の米国法典第35編120条に基づいて下記の米国特許出願に記載された権利、又は米国を指定している特許協力条約365条 (c) に基づく権利をここに主張します。また、本出願の各請求範囲の内容が米国法典第35編112条第1項又は特許協力条約で規定された方法で先行する米国特許出願に開示されていない限り、その先行米国出願書提出日以降で本出願書の日本国内または特許協力条約国際提出日までの期間中に入手された、連邦規則法典第37編1条56項で定義された特許資格の有無に関する重要な情報について開示義務があることを認識しています。

(Application No.) (出願番号)	(Filing Date) (出願日)	(Status: Patented, Pending, Abandoned) (現況: 特許可済、係属中、放棄済)
(Application No.) (出願番号)	(Filing Date) (出願日)	(Status: Patented, Pending, Abandoned) (現況: 特許可済、係属中、放棄済)

私は、私自信の知識に基づいて本宣言書中で私が行なう表明が真実であり、かつ私の入手した情報と私の信じるところに基づく表明が全て真実であると信じていること、さらに故意になされた虚偽の表明及びそれと同等の行為は米国法典第18編第1001条に基づき、罰金または拘禁、もしくはその両方により処罰されること、そしてそのような故意による虚偽の声明を行なえば、出願した、又は既に許可された特許の有効性が失われることを認識し、よってここに上記のことく宣誓を致します。

I hereby claim foreign priority under Title 35, United States Code, Section 119 (a)-(d) or 365(b) of any foreign application(s) for patent or inventor's certificate, or Section 365(a) of any PCT International application which designated at least one country other than the United States, listed below and have also identified below, by checking the box, any foreign application for patent or inventor's certificate, or PCT International application having a filing date before that of the application on which priority is claimed.

<input type="checkbox"/>	<input type="checkbox"/>
<input type="checkbox"/>	<input type="checkbox"/>

I hereby claim the benefit under Title 35, United States Code, Section 119(e) of any United States provisional application(s) listed below.

(Application No.) (出願番号)	(Filing Date) (出願日)
-----------------------------	------------------------

I hereby claim the benefit under Title 35, United States Code, Section 120 of any United States application(s), or Section 365(c) of any PCT International application designating the United States, listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States or PCT International application in the manner provided by the first paragraph of Title 35, United States Code Section 112, I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, Section 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of application.

(Application No.) (出願番号)	(Filing Date) (出願日)	(Status: Patented, Pending, Abandoned) (現況: 特許可済、係属中、放棄済)
(Application No.) (出願番号)	(Filing Date) (出願日)	(Status: Patented, Pending, Abandoned) (現況: 特許可済、係属中、放棄済)

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

## Japanese Language Declaration

(日本語宣言書)

委任状：私は下記の発明者として、本出願に関する一切の手続きを米特許商標局に対して遂行する弁理士または代理人として、下記の者を指名いたします。  
(弁理士、または代理人の指名及び登録番号を明記のこと)

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith: (list name and registration number)



022850

書類送付先

Send Correspondence to:



022850

直接電話連絡先：(名前及び電話番号)

Direct Telephone Calls to: (name and telephone number)

(703) 413-3000

■ 単独発明者または第一の共同発明者の氏名 <u>1-00</u>		Full name of sole or first joint inventor <u>Osamu HIROI</u>	
■ 発明者の署名	日付	Inventor's signature <u>Osamu Hiroi</u>	Date 12/6/2001
■ 住所		Residence <u>TOKYO 100-8310 JAPAN JPY</u>	
■ 国籍		Citizenship <u>Japanese</u>	
郵便の宛先		Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN	
第二の共同発明者の氏名 <u>2-00</u>		Full name of second joint inventor, if any <u>Jun ARAGANE</u>	
第二の共同発明者の署名	日付	Second joint Inventor's signature <u>Jun Aragane</u>	Date 12/6/2001
住所		Residence <u>TOKYO 100-8310 JAPAN JPY</u>	
国籍		Citizenship <u>Japanese</u>	
郵便の宛先		Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN	

(第三以降の共同発明者についても同様に記載し、署名すること)  
(Supply similar information and signature for third and subsequent joint inventors.)

## Japanese Language Declaration

(日本語宣言書)

第三の共同発明者の氏名	3-00			Full name of third joint inventor, if any <u>Shoiji YOSHIOKA</u>
第三の共同発明者の署名	日付	Third joint Inventor's signature <u>Shoiji YOSHIOKA</u> Date 12/6/2001		
住所				Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍				Citizenship Japanese ✓
郵便の宛先				Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

第四の共同発明者の氏名	4-00			Full name of fourth joint inventor, if any <u>Yasuhiro YOSHIDA</u>
第四の共同発明者の署名	日付	Fourth joint Inventor's signature <u>Yasuhiro YOSHIDA</u> Date 12/6/2001		
住所				Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍				Citizenship Japanese ✓
郵便の宛先				Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

第五の共同発明者の氏名	5-00			Full name of fifth joint inventor, if any <u>Tetsuyuki KURATA</u>
第五の共同発明者の署名	日付	Fifth joint Inventor's signature <u>Tetsuyuki KURATA</u> Date 12/6/2001		
住所				Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍				Citizenship Japanese ✓
郵便の宛先				Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

第六の共同発明者の氏名	6-00			Full name of sixth joint inventor, if any <u>Yukiyasu NAKAO</u>
第六の共同発明者の署名	日付	Sixth joint Inventor's signature <u>Yukiyasu NAKAO</u> Date 12/6/2001		
住所				Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍				Citizenship Japanese ✓
郵便の宛先				Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

(第六またはそれ以降の共同発明者に対しても同様な情報および署名を提供すること。)

(Supply similar information and signature for third and subsequent joint inventors.)

TOS0014100301

第七の共同発明者の氏名	Full name of seventh joint inventor, if any		
第七の共同発明者の署名	日付	Hiroaki URUSHIBATA	Seventh joint Inventor's signature Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post Office Address		
第八の共同発明者の氏名	Full name of eighth joint inventor, if any		
第八の共同発明者の署名	日付	Yoshio KASUGA	Eighth joint Inventor's signature Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post Office Address		
第九の共同発明者の氏名	Full name of ninth joint inventor, if any		
第九の共同発明者の署名	日付	Seiichi MIMURA	Ninth joint Inventor's signature Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post Office Address		
第十の共同発明者の氏名	Full name of tenth joint inventor, if any		
第十の共同発明者の署名	日付	Hironori OZAKI	Tenth joint Inventor's signature Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post Office Address		

00000000000000000000000000000000

第十一の共同発明者の氏名 <i>11-00</i>	Full name of eleventh joint inventor, if any <u>Hideo ICHIMURA</u>
第十一の共同発明者の署名 日付	Eleventh joint Inventor's signature <u>Hideo Ichimura</u> Date 12/6/2001
住所	Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍	Citizenship Japanese
郵便の宛先	Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN
第十二の共同発明者の氏名 <i>12-00</i>	Full name of twelfth joint inventor, if any <u>Daigo TAKEMURA</u>
第十二の共同発明者の署名 日付	Twelfth joint Inventor's signature <u>Daigo Takemura</u> Date 12/6/2001
住所	Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍	Citizenship Japanese
郵便の宛先	Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN
第十三の共同発明者の氏名 <i>13-00</i>	Full name of thirteenth joint inventor, if any <u>Hisashi SHIOTA</u>
第十三の共同発明者の署名 日付	Thirteenth joint Inventor's signature <u>Hisashi Shiota</u> Date 12/6/2001
住所	Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍	Citizenship Japanese
郵便の宛先	Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN
第十四の共同発明者の氏名 <i>14-00</i>	Full name of fourteenth joint inventor, if any <u>Shigeru AIHARA</u>
第十四の共同発明者の署名 日付	Fourteenth joint Inventor's signature <u>Shigeru Aihara</u> Date 12/6/2001
住所	Residence <u>TOKYO 100-8310 JAPAN JPX</u>
国籍	Citizenship Japanese
郵便の宛先	Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

第十五の共同発明者の氏名 15-00	Full name of fifteenth joint inventor, if any <u>Takashi NISHIMURA</u>
第十五の共同発明者の署名 日付	Fifteenth joint Inventor's signature <u>Takashi Nishimura</u> Date 12/6/2001
住所	Residence <u>TOKYO</u> 100-8310 JAPAN <u>JPX</u>
国籍	Citizenship Japanese
郵便の宛先	Post Office Address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi 2-chome, Chiyoda-ku, TOKYO 100-8310 JAPAN

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